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Methoxypropyl-Acetate

Description

Medium-volatility solvent with a mild odour. Does not contain any hydroxyl groups and has good solvent power for numerous resins and dyes. Its main applications are in the coatings and printing inks, where it can replace ethoxyethyl acetate in many cases.

Properties

Methoxypropyl-Acetate is a clear, slightly hygroscopic liquid with a mild odour. It is freely miscible with most common organic solvents, but has only limited miscibility with water. By virtue of its ether and ester groups, Methoxypropyl-Acetate enters into reactions that are characteristic of ethers and esters and display their solvent power. For instance, it dissolves numerous natural and synthetic resins, waxes, fats and oils. Since Methoxypropyl-Acetate may react with the oxygen in the air to form peroxides, Novonol supplies it inhibited with butylated hydroxytoluene (BHT).

Chemical Nature

1-Methoxy-2-propyl acetate, Propylene glycol monomethyl ether-1, 2-acetate

 $\begin{array}{lll} \text{Molecular formula} & C_6 \text{H}_{12} \text{O}_3 \\ \text{Molar mass} & 132.16 \text{ g/mol} \\ \text{CAS number} & 108-65-6 \\ \text{EC number} & 203-603-9 \end{array}$

Spezfikationen / Specifications

| Property | Test Method | Unit | Methoxypropyl- Acetate |
|--|---|--------|---------------------------|
| Density @ 20°C | DIN 51757 | kg/l | 0,965 – 0,970 |
| Methoxypropyl-Acetate* Methoxypropanol | | % % | ≥ 99,5 ≤ 0,3 |
| Acetic Acid Water | DIN EN ISO 2114 DIN EN 51777, part 1 | % | ≤ 0,03 ≤ 0,05 |
| Color Value (Hazen) | DIN EN ISO 6271 | | ≤ 10 |

^{*}Methoxypropyl Acetate contains less than 0.3 % of the 2-methoxy-1-propyl acetate isomer.

The abovementioned data shall constitute the agreed contractual quality of the product at the time of passing risk. The data are controlled at regular intervals as of our quality assurance program. Neither these data nor the properties of the product specimens shall imply any legally binding guarantee of certain properties or of fitness for a specific purpose. No liability of ours can be derived therefrom.